

# IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Reissue application of

Kinam Park et al.

27

U.S. Serial No: 10/807,227

: Art Unit: 1711

Filed: March 22, 2004

Examiner: John M. Cooney

For: HYDROGEL COMPOSITES AND

SUPERPOROUS HYDROGEL COMPOSITES :

HAVING FAST SWELLING, HIGH

Confirmation No: 1689

MECHANICAL STRENGTH, AND SUPERABSORBENT PROPERTIES

# DECLARATIONS UNDER 37 CFR 1.131

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

#### Sir:

We, Kinam Park, Jun Chen and Haesun Park hereby declare the following:

- We are the co-inventors of claims 1-40 in the aboveidentified patent application.
- 2. We did conceive and reduce to practice the claimed invention prior to May 7, 1997. The application directed to this invention was filed on May 13, 1997.
- 3. In support of our Declarations we refer to the attached Statement of Dr. James H. Meadows, which establishes that the instant invention disclosure was sent by Federal Express to the predecessor law firm, Lowe,

U.S. Serial No. 10/936,613

Price, LeBlanc & Becker, on December 16, 1996, as is evidenced by a facsimile letter signed by Dr. Kinam Park, to which is applied a "Received" date stamp of December 16, 1996.

- 4. The Statement of Dr. Meadows also provides a copy of a letter from him addressed to Dr. Kinam Park dated May 2, 1997, which forwards a final draft of the application for review.
- 5. The aforementioned documents prove that the present invention was conceived and reduced to practice well before May 7, 1997.

We further declare that all statements made herein of our own knowledge are true and that all statements made on information and belief are true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

7/11/06
Date

Date

Jun Chen

All John Park

Maeau Park

Haesun Park

PATENT

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Date	Kinam Park	
July 12, 2006	Jun Chen	
Date	Haesun Park	



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## STATEMENT OF JAMES H. MEADOWS

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

- I, James H. Meadows, am the attorney representing applicants in the above-identified patent application. I hereby state the following:
  - 1. The instant invention disclosure was sent by Federal Express to the predecessor law firm, Lowe, Price, LeBlanc & Becker, on December 16, 1996, as is evidenced by a facsimile letter signed by Dr. Kinam Park, to which is applied a "Received" date stamp of December 16, 1996.
  - A final draft of the application was sent by me to Dr.
     Kinam Park for review on May 2, 1997.

## U.S. Serial No. 10/807,227

3. Copies of the aforementioned documents are forwarded herewith.

I further declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

6/11/06 Date A Meadows Ph.D.

Attachments



# PURDUE UNIVERSITY

School of Pharmacy, West Lafayette, IN 47907-1336

Tel: (317) 494-7759

Fax: (317) 496-1903

E-mail: esp@omni.cc.purdue.edu

Kinam Park, Ph.D. Professor of Pharmaceutics

RECEIVED

December 16, 1996

Mr. Robert L. Price
Law Offices
Lowe, Price, Leblanc & BeckeRFRFIVED VIA FACSIMILE
Alexandria, VA 22314

Tel: 703-684-1111

Fax: 703-684-1124 (or 1145)

Re: Disclosure on "Synthesis of superporous hydrogel composites with fast swelling, high mechanical strength, and superabsorbent properties" by Kinam Park, Iun Chen, and Haesun Park

Dear Mr. Price:

I sent the above mentioned disclosure to you by Federal Express. Please fax me a note upon your receipt.

Thank you very much for your help and look forward to working with you.

With best regards.

Kinam Park

# LOWE PRICE LEBLANC & BECKER

SUITE 300 99 CANAL CENTER PLAZA ALEXANDRIA, VIRGINIA 22314

TELEPHONE: 703-684-1111 FACSIMILE: 703-684-1124

PATENT TRADEMARK COPYRIGHT UNFAIR COMPETITION LAW LITIGATION

STEICON VALLEY, CA FAIRMONT PLAZA SUITE 330 50 WEST SAN FERNANDO STREET SAN JOSE, CALIFORNIA 95113 TELEPHONE. 408-271-2300 FACSIMILE, 408-271-2310

DEMETRA J. MILLS WILLIAM H. BEHA ERIC J. KRAUS\* LEON R. TURKEVICH\* KAREN C. MAHER\* MICHAEL E. MCCABE. JR. . ROBERT S. GREEN GREG P. ROGGIN\* EDWARD A. BECKER\* KENNETH M. BERNER\* JAMES H. MEADOWS\* STEPHEN C. CARLSON MARCEL K. BINGHAM\* GLENN SNYDER\* THOMAS A. JOLLY\* CARL L. BRANDT\*

OF COUNSEL CHITTARANIAN N. NIRMEL\* HENRY SHUR IRVING R. PELLMAN\* MICHAEL G. GILMAN CHRISTOPHER J. PALERMO\*

PATENT AGENTS ALEXANDER V. YAMPOLSKY KEITH J. TOWNSEND

VIA COURIER

214 5673 923

May 2, 1997

Dr. Kinam Park Professor of Pharmaceutics School of Pharmacy PURDUE UNIVERSITY West Lafayette, IN 47907-1336 USA

New U. S. Patent Application

HYDROGEL COMPOSITES AND SUPERPOROUS HYDROGEL COMPOSITES HAVING FAST SWELLING, HIGH MECHANICAL

STRENGTH, AND SUPERABSORBANT PROPERTIES

Inventor: Kinam Park et al. Our Reference: 3178-001

Dear Dr. Park:

Re:

, , , ,

ALLAN M. LOWE\*

ROBERT L. PRICE

STEPHEN A. BECKER

ROBERT E. LEBLANC

KENNETH E. KROSIN

CHARLES D. OSSOLA

GENE Z. RUBINSON\*

ARTHUR J. STEINER\* EDWARD J. WISE

DAVID L. STEWART

JOHN A. HANKINS\*

IRAH H. DONNER\*

JUDITH SAPP"

BRIAN D. HICKMAN\*

TIMOTHY R. DEWITT

THAN MEMBERSHIP OTHER THAN VINCINIA

KEITH E. GEORGE

BENJAMIN J HAUPTMAN

GILBERTO M. VILLACORTA\*

CHRISTOPHER W. BRODY

ALFRED A. STADNICKI\*

ISRAEL GOPSTEIN\*

FRANK P. PRESTA

Please find enclosed a draft of the above-mentioned patent application which incorporates the changes forwarded with your April 21, 1997 letter. As you will note, this draft reorganizes and revises the claims significantly from the previous draft which you considered. Consistent with our discussions on May 1, 1997, you will note that the claims now are directed to hydrogel composites and superporous hydrogel composites, as well as methods of making these materials.

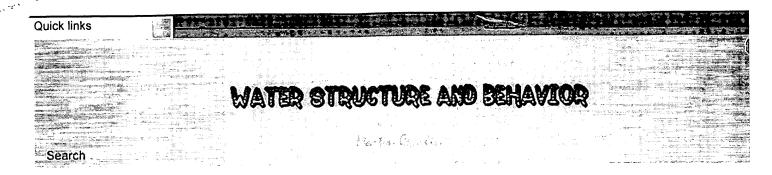
Please review the draft at your earliest convenience and provide me with your comments. If you should have any further questions, please do not hesitate to contact me.

Very truly yours,

LOWE PRICE LEBLANC & BECKER H Meadows

James H. Meadows, Ph.D.

Enclosure



# Carboxymethylcellulose (CMC)

#### Source

Carboxymethylcellulose (CMC; E466) is a derivative of cellulose formed by its reaction with alkali and chloroacetic acid.

#### Structural unit

The CMC structure is based on the  $\beta$ -(1 $\rightarrow$ 4)-D-glucopyranose polymer of cellulose. Different preparations may have different degrees of substitution, but it is generally in the range 0.6 - 0.95 derivatives per monomer unit.

#### Molecular structure

CMC molecules are somewhat shorter, on average, than native cellulose with uneven derivatization g areas of high and low substitution. This substitution is mostly 2-O- and 6-O-linked, followed in order importance by 2,6-di-O- then 3-O-, 3,6-di-O-, 2,3-di-O- lastly 2,3,6-tri-O-.linked. It appears that the substitution process is a slightly cooperative (within residues) rather than random process giving sligh higher than expected unsubstituted and trisubstituted areas. CMC molecules are most extended (rodat low concentrations but at higher concentrations the molecules overlap and coil up and then, at hig concentrations, entangle to become a thermoreversible gel. Increasing ionic strength and reducing pl both decrease the viscosity as they cause the polymer to become more coiled.

### **Functionality**

CMC dissolves rapidly in cold water and mainly used for controlling viscosity without gelling (CMC, at typical concentrations, does not gel even in the presence of calcium ions). As its viscosity drops durir heating, it may be used to improve the volume yield during baking by encouraging gas bubble formation. Its control of viscosity allows use as thickener, phase and emulsion stabilizer (*e.g.* with mi casein), and suspending agent. CMC can be also used for its water-holding capacity as this is high ev at low viscosity; particularly when used as the Ca<sup>2+</sup> salt. Thus, it is used for retarding staling and reducing fat uptake into fried foods.

The average chain length and degree of substitution are of great importance; the more-hydrophobic lower substituted CMCs are thixotropic but more-extended higher substituted CMCs are pseudoplastic low pH, CMC may form cross-links through lactonization between carboxylic acid and free hydroxyl groups.

The solution properties of a range of commercial CMC's have been investigated [879].

Interactive structures are available (COW [Plug-in, ActiveX], 20 KB; Chime, 5 KB).

Please submit any comments and suggestions you may have.

Site Index | Hydrocolloids | Polysaccharide hydration | Hydrogen bonding



This page was last updated by Martin Chaplin on 19 April, 2006